Duration: 3 hours Total Marks: 80

- 1. Question number one is compulsory.
- 2. Attempt any three of the remaining questions.
- 3. Each question carries equal marks.
- 4. Figures to the right indicate marks.
- 5. Make suitable assumptions when required.
- 01 Answer any four of the following Explain TSA for regeneration of adsorbents (a) Give classification of membranes. **(b)** (c) Explain working principle of supercritical extraction Describe molecular distillation (e) What is the principle of operation of ultrafiltration Preparative chromatography **(f)** An industrial wastewater contains 100 mg/L chlorophenol and is going to be $\mathbf{Q2}$ (a) treated by carbon adsorption. 99% removal is desired. The wastewater is discharged at a rate of 4 x 10^5 L/day. If Freundlich isotherm (q=6.75 $C^{0.41}$, where q is in mg/g) is used, calculate The carbon requirement for:-(1) A single, mixed contactor (ii) Two mixed contactors in series with intermediate concentration of 50 mg/L10 **(b)** Explain design of fixed bed adsorption column for separation With the help of neat sketches and examples explain reactive distillation. 10 (a) Illustrate commercial application of supercritical extraction **(b) 10** Explain any two membrane modules in detail. 10 (a) Explain concentration polarization. 10 A cellulose acetate membrane is used for desalination RO system. Pure water 10 flux is $100(\frac{L}{m^2}h)$ at hydrostatic pressure differential of 100 bar. A 4% NaCl solution is to be desalinated at an operating pressure difference of 90bar. Salt rejection is 98%. Osmotic pressure of 3.5% NaCl solution is 17.62bar. The polarization modulus is unity. Calculate the water and salt flux. Explain in detail PSA technology. **10 (b)** Derive the relation for liquid permeation process in 'Dialysis' i.e. $J_{AS} = \frac{C_1 - C_2}{\left(\frac{1}{kC_1}\right) + \left(\frac{1}{kC_2}\right) + \left(\frac{1}{P_m}\right)}$ (a) 10

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Explain Liquid liquid Chromatographic process