Duratio	n: 3 Hours Total marks: 100	
Instruct	tions to the candidates:	
1) All t	the questions are compulsory. Choice is internal.	6
2) Figu	res to the right indicate full marks.	20
3) All (questions carry equal marks.	
4) Drav	w flowcharts /diagrams wherever necessary.	p &
5) Use o	of simple calculator is allowed.	
Q1A)	Fill in the blanks (any three):	3
i)	Ketone bodies include	20
ii)	Enzymes for the fatty acid oxidation are present in the	2/2/2
iii)	is the end product of beta oxidation of even carbon saturated fatty	50 ^T
• `	acids.	
iv)	Acetyl CoA carboxylase adds a group to acetyl CoA.	
v)	Synthesis of fatty acid is termed as	
O1D)		2
Q1B)	Answer in brief any one of the following:	3
i)	Discuss the contribution of Knoop to the field of lipid metabolism.	
ii)	Briefly explain the fate of glycerol obtained on breakdown of triglycerides.	
010)		,
Q1C)		6
i)	With the help of a schematic representation, explain how the carnitine shuttle	
	plays an important role in determining the rate of beta oxidation. Emphasize on its significance in lipolysis.	
ii)	Explain in detail the structure and function of the multi-enzyme complex involved	
)	in synthesis of fatty acids.	
Q1D)	Answer any one of the following:	8
i)	With the help of structures and enzymes involved, explain the oxidation of	
	Eicosanoic acid (C-20). Also, elaborate on the bioenergetics of the same.	
ii)	'Diabetic individual usually excrete ketone bodies in their urine.' Elaborate in detail	
55.57	on the formation and utilization of ketone bodies by a diabetic individual.	
		2
Q2A)	· & · & · & · & · & · & · & · & · & · &	3
i)		
ii)	is regenerated at the end of urea cycle.	
iii)	functions as the secondary messenger.	
iv)	Congenital hypothyroidism is also known as	
(v)	hormone causes suppression of immune system.	
Q2B)	Answer in brief any one of the following:	3
(Q2B)	8 V - S S V - S V	5
2 % -0, 4	Explain the sequential process of insulin biosynthesis	
V V (VV)	The state of the s	

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Q2C)	Differentiate between <u>any one</u> of the following:	6
i)	Type I and Type II diabetes mellitus	
ii)	Oxidative and non-oxidative deamination	56
Q2D)	Write detailed answers to any one of the following:	8
i)	Discuss in detail thyroid hormones under the following headings: (a) chemistry (b) biosynthesis (c) physiological effects.	000
ii)	With the help of structures, explain the final pathway of protein degradation. Also, mention the cellular location and trace the origin of the two nitrogen atoms.	\$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$
Q3A)	Fill in the blanks (any three):	3
i)	wavelength range is associated with visible colorimeter.	7/2
ii)	Desk top centrifuge has a capacity of ml.	07
iii)	The first fraction obtained at 600 rpm in differential centrifugation of liver homogenisation is	
iv)	Unit of sedimentation coefficient is	
v)	Rate of sedimentation is proportional to viscosity of a medium.	
Q3B)	Attempt in brief any one:	3
i)	Explain the basic principle of sedimentation.	
ii)	A series of 3 colored glass plates of equal thickness are placed in a light beam. Each sheet absorbs one quarter of the light incident upon it. What is the intensity of the light transmitted by the third glass plate?	
Q3C)	Attempt any one of the following:	6
i)	Compare and contrast: Rate zonal and isopycnic centrifugation. Name any two examples of compounds used as density gradient materials.	
ii)	State Beer-Lambert's law and add a note in detail about deviation and limitation of the same.	
Q3D)	Write detailed answers of any one of the following:	8
	Write a detailed note on analytical ultracentrifugation.	
/_ Y V 7	Explain construction of spectrophotometer with respect to following a) Source of light b) wavelength selector	
	c) sample containers d) detector	
Q4A)	Fill in the blanks (any three):	3
	dye is used to stain nucleic acids.	
ii)	High molecular weight particles travel in a supporting medium with low porosity than low molecular weight components.	
iii)	Agarose gel has molecular sieving effect.	
iv)	SDS is a/an detergent.	
	Carbohydrates can be separated by electrophoresis using buffer.	

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Q4B) i)	Answer in brief <u>any one</u> of the following: Give a brief account on Native PAGE.	3
ii)	An inert supporting medium is chosen for electrophoresis. How does it influence the movement of sample?	
Q4C)	Attempt any one of the following:	6
i)	Explain the effect of 1) composition 2) ionic strength and 3) pH of the buffer on the rate of migration of sample in electrophoresis.	A A
ii)	Elaborate on the characteristics and applications of 1) paper and 2) cellulose acetate as supporting media.	
Q4D)	Answer any one of the following:	8
i)	State True or False, giving reason: Pore size can be controlled in polyacrylamide gels. With the help of reaction explain the formation of the gel including the role of ammonium persulphate and TEMED.	20
ii)	Justify: Electrophoresis has wide array of applications.	
Q5A)	Answer any four of the following:	16
i)	Discuss the formation of malonyl CoA. Elaborate on its significance to lipid metabolism	10
:)	OR Describe the pathophysiological conditions leading to ketone bodies and explain	
i)	their effect on human physiology.	
ii)	Explain the physiological functions of glucocorticoids. OR	
ii)	In detail, elaborate on the reaction mechanism of transamination reactions.	
iii)	Write a note on general properties of gradient material. OR	
iii)	How would you estimate the unknown concentration of protein using colorimetric method?	
iv)	Briefly explain the working of type of electrophoresis that uses two different gels at a time.	
	OR	
iv)	Describe the principle of electrophoresis. Give any one example each of a buffer and a dye used for separation of protein.	
Q5B)	State true or false (any four):	4
i)	Lambda max is independent of concentration for a particular biomolecule	
ii)	Svedberg (S) is the unit in centrifugation.	
iii)	The coenzyme required for oxidative deamination is pyridoxal phosphate.	
iv)	Fatty acid synthase complex has a structural and functional division.	
v)	Intact protein molecules are separated using SDS-PAGE.	
vi)	pH of the stacking gel is 2 units higher than that of separating gel. ***********************************	

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