		T. E. sean-VI (CBSGS) cheanical- MTO-II Mass Transfer Devation-II QP CODE:	29/11/16
		Mass Transfer - Peration - II QP CODE:	574203
		[3 Hours]	[Total Marks: 80]
		(1) Q.1 is compulsory. (2) Attempt any 3 from the remaining	
		(2) Attempt any 3 from the remaining 5 questions.(3) Use graph paper, if required.	
	Q.1	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	[05]
		(b) Compare beween extraction and distillation as seperation methods.	[05]
		(c) Explain pervaporation.	[05]
		(d) Write a short note on Swanson Walker crystallizer.	[05]
	Q.2	(a) Explain Ponchon- savarit method for multi stage tray towers.	[12]
		(b) Derive Fenske's equation for minimum no. of stages at total reflux.	[08]
			1677
	Q.3	(a) Water - dioxane solution forms a minimum boiling azeotrope at atmospheric pro-	ressure and
		cannot be separated by ordinary distillation methods. Benzene forms no azeotro	
		dioxane and may be used as an extraction solvent. At 25°C the equilibrium distr	ibution
		of dioxane between water and benzene is as follows:	
		wt % in water 5.1 18.9 25.2	
		wt % in benzene 5.2 22.5 32	
		G) Calculate have a constant of the constant o	
		(i) Calculate benzene requirement for single stage operation.(ii) If the extraction is done in counter current fashion, what is the minimum so	[80]
		requirement in kg/hr?	[06]
		(b) Write a short note on any one liquid - liquid extraction equipment.	[06]
2	Q. 4	(a) 360 kg/hr of halibut liver is to be extracted in a counter current cascade with ether	r to recover oil.
		The other which has been used partially contains 2.5 % oil. The fresh liver contains	s 25% oil and
		is to be extracted to composition 2% oil. (On solvent free basis) 250 kg of solvent is	s to be used.
		(i) What % of oil entering with the liver is recovered in the extract?	[08]
		(ii) How many equilibrium stages are required?	[07]

[TURN OVER

The data are as given below:

Concentration solution)	(kg	oil/kg	0	0.1	0.2	0.3	0.4	0.5
Kg solution/kg ex	hausted	liver	0.288	0.368	0.44	0.51	0.6	0.71

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	(b) Explain the construction and working of Bollmann extractor.	[05]
Q.5.	(a) Explain desired properties of adsorbents.	[06]
	(b) Write the industrial applications of adsorption. Describe any 4 adsorbents.	[08]
	(c) Write a short note on steam distillation.	[06]
Q. 6.	Write short notes on the following:	
	(a) Azeotropic distillaiton	[05]
	(b) Binodal solubility curve	[05]
	(c) Ion exchange process	[05]
	(d) Electrodialysis	[05]