Paper / Subject Code: 50702 / Engineering Chemistry-I

Time: 3 Hours

18-Nov-2019 1T00523 - S.E.(Chemical Engineering)(SEM-III)(Choice Base) / 50702 - Engineering Chemistry-I 77006

Marks: 80

		Please check whether you have got the right question paper.	355
N.B.:	1. Qu	estions no. 1 is compulsory.	
	2. Att	empt any three questions from remaining five questions.	30 30 30 30 30 30 30 30 30 30 30 30 30 3
Q.1	Attempt any four questions of the following:		[20]
	a)	Explain Geometry & hybridization of BrF ₃ molecule on the basis of VSEPR	300
		theory.	
	b)	Discuss preparation, bonding, hybridization and geometry of Fe ₂ (CO)9 molecule.	5 6 V
	c)	Differentiate between Kinetically controlled and thermodynamically controlled	, ox
		reaction.	
	d)	Discuss structure and stability of carbanion.	
	e)	Differentiate between Photochemical & Thermochemical reactions.	
	f)	Give Nomenclature for following.	
		i) Na ₂ [Zn(OH) ₄] ii) [Co(NH ₃) ₆] Cl ₃	
Q.2	a)	Write note on Effective atomic number (EAN). Explain with an example.	[05]
	b)	Draw Molecular Orbital Diagram of F ₂ molecule. Comment on Bond order and	
		Magnetic character.	[05]
	c)	Write note on Iron containing Protein.	[05]
	d)	Explain Quantum yield. Give reasons for high Quantum yield.	[05]
	E FER		
Q.3	() (a)	Discuss Sulphonation of naphthalene as kinetically controlled and	
		thermodynamically controlled reaction.	[05]
CON TO	(b)	Explain S-P and P-P orbitals overlapping for formation of σ & σ* by LCAO	
STAN		method.	[05]
	c)	Discuss Reimer-Tiemann reaction with its applications.	[05]
	d)	Discuss Biochemistry of Zinc containing metalloprotein.	[05]

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Q.4	a)	Calculate CFSE for d ⁵ and d ⁷ system in strong field and weak field for octahedral	S. S. L.
		complexes.	[05]
	b)	Explain Wohl-Ziegler reaction.	[05]
	c)	What is hydrogen bonding? Explain Intra molecular and inter molecular hydrogen	300
		bonding with an examples.	[05]
	d)	Discuss Jablonski diagram.	[05]
Q.5	a)	Explain Pinacol - Pinacolone rearrangement reaction.	[05]
	b)	Explain geometrical isomerism with example in co-ordination compounds with	
		respect to CN-4.	[05]
	c)	Define and explain Intermediate and Transition state with suitable example.	[05]
	d)	Differentiate between Fluorescence & Phosphorescence.	[05]
Q.6	a)	Calculate EAN for following Compound	
		i) [Pt (Cl) ₄ (H ₂ O) ₂] ii) K ₄ [Fe (CN) ₆].	[05]
	b)	Discuss the formation and structure of carbocation.	[05]
	c)	Explain oxygen transportation in hemoglobin.	[05]
	d)	Discuss the VBT and give the limitations of VBT.	[05]
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